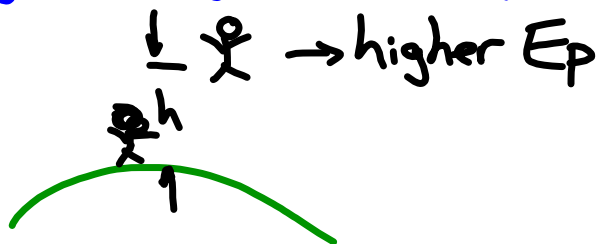


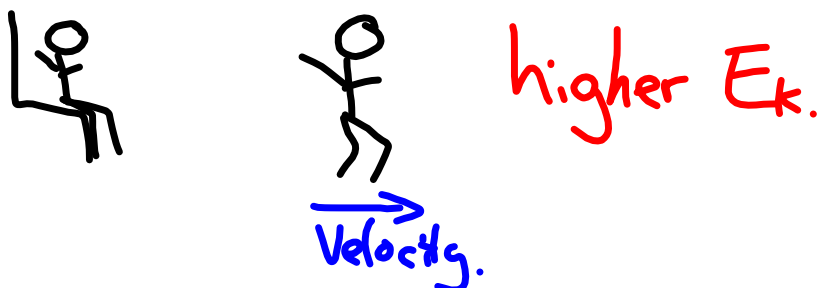
Mechanical Energy

Definition - Energy of position or motion

Gravitational Potential Energy - E_p
- energy possessed by an object due to its height in a gravitational field.



Kinetic Energy - E_k
- energy possessed by an object due to its relative motion compared to a second object.



Examples

A. Energy of motion : What is the work required (i.e. the change in energy) to accelerate a 10kg object from 0 to 30m/s in 20s?

$$\begin{aligned} W &= F \cdot d \\ &= 15\text{N} \cdot 300\text{m} \\ &= 4500\text{J} \end{aligned}$$

$$\begin{aligned} F &= ma \\ &= m \left(\frac{v_2 - v_1}{\Delta t} \right) \\ &= 10 \left(\frac{30 - 0}{20} \right) \text{②} \\ &= 15\text{N} \\ d &= \left(\frac{v_1 + v_2}{2} \right) \Delta t \text{ ①} \\ &= \left(\frac{0 + 30}{2} \right) 20 \\ &= 300\text{m} \end{aligned}$$

Predict - What if we wanted to accelerate the same object as in **Part A** from 0 to 30m/s, but this time in 5s? Would the energy requirement be higher, lower or the same?

$$\begin{aligned} W &= F \cdot d \\ &= 60\text{N} \cdot 75\text{m} \\ &= 4500\text{J} \end{aligned}$$

$$\begin{aligned} F &= ma \\ &= m \left(\frac{v_2 - v_1}{\Delta t} \right) = 60\text{N} \\ d &= \left(\frac{v_1 + v_2}{2} \right) \Delta t = 75\text{m} \end{aligned}$$

Standard Formula for Kinetic Energy

Calculate the work required to accelerate an object of mass, m , from rest to velocity V (m/s) in t seconds.



$$W = F \cdot d$$

$$F = ma$$

$$= m \left(\frac{v_2 - v_1}{\Delta t} \right)$$

$$= m \left(\frac{V - 0}{t} \right)$$

$$= \frac{mV}{t}$$

$$d = \left(\frac{v_1 + v_2}{2} \right) \Delta t$$

$$= \left(\frac{0 + V}{2} \right) t$$

$$= \frac{Vt}{2}$$

$$W = \left(\frac{mV}{\cancel{t}} \right) \left(\frac{V\cancel{t}}{2} \right)$$

$$= \frac{mV^2}{2}$$

*

$$E_k = \frac{1}{2} mV^2$$

called kinetic energy

- energy a moving object has relative to not moving.

Examples:

B. Calculate the kinetic energy of an electron ($m=9.11 \times 10^{-31}$ kg) moving at 80% of the speed of light ($c=3.0 \times 10^8$ m/s)

$$E_k = \frac{1}{2}mv^2$$

$$= \frac{1}{2}(9.11 \times 10^{-31} \text{ kg})(2.4 \times 10^8 \text{ m/s})^2$$

$$= 2.6 \times 10^{-14} \text{ J}$$

$$v = 0.80 \times 3.0 \times 10^8 \text{ m/s}$$

$$= 2.4 \times 10^8 \text{ m/s}$$

$$\text{kg m}^2/\text{s}^2 = \text{J}$$

C. Calculate the kinetic energy of 1500kg car moving at 50 km/hr.

$$E_k = \frac{1}{2}mv^2$$

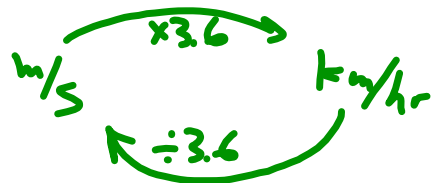
$$= \frac{1}{2}(1500 \text{ kg})(13.89 \text{ m/s})^2$$

$$= 1.4 \times 10^5 \text{ J} \checkmark$$

~~$$144676 \text{ J}$$~~

$$145000 \text{ J} \checkmark$$

$$140000 \text{ J} \checkmark$$

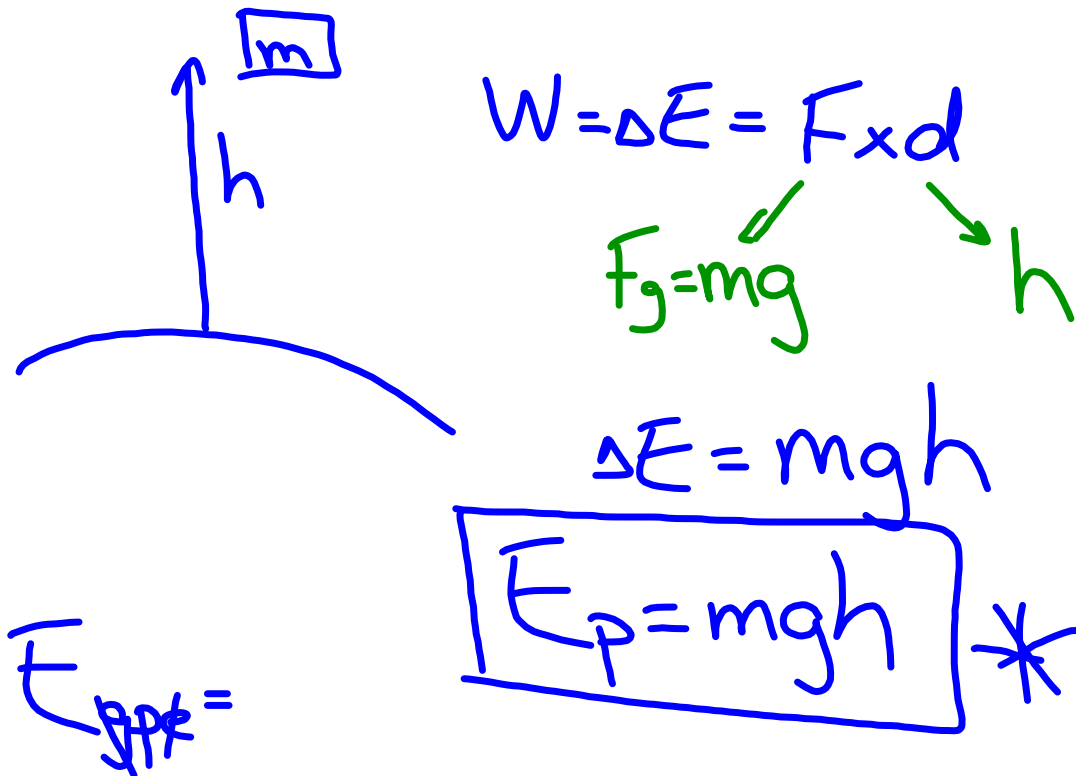


$$50 \text{ km/hr} = 50 \div 3.6 \text{ m/s}$$

$$= 13.89 \text{ m/s}$$

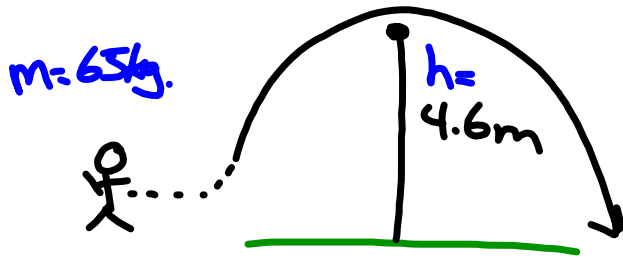
Gravitational Potential Energy

energy an object has due to its height above the ground.



Example

A. Pole Vaulting : A 65 kg pole vaulter can clear a bar that is 4.6m above the track. What is the potential energy of the pole vaulter at the top of the jump?



$$\begin{aligned} E_p &= mgh \\ &= 65\text{kg} \times 9.8\frac{\text{m}}{\text{s}^2} \times 4.6\text{m} \\ &= 2900\text{J} \end{aligned}$$

Conservation of Energy

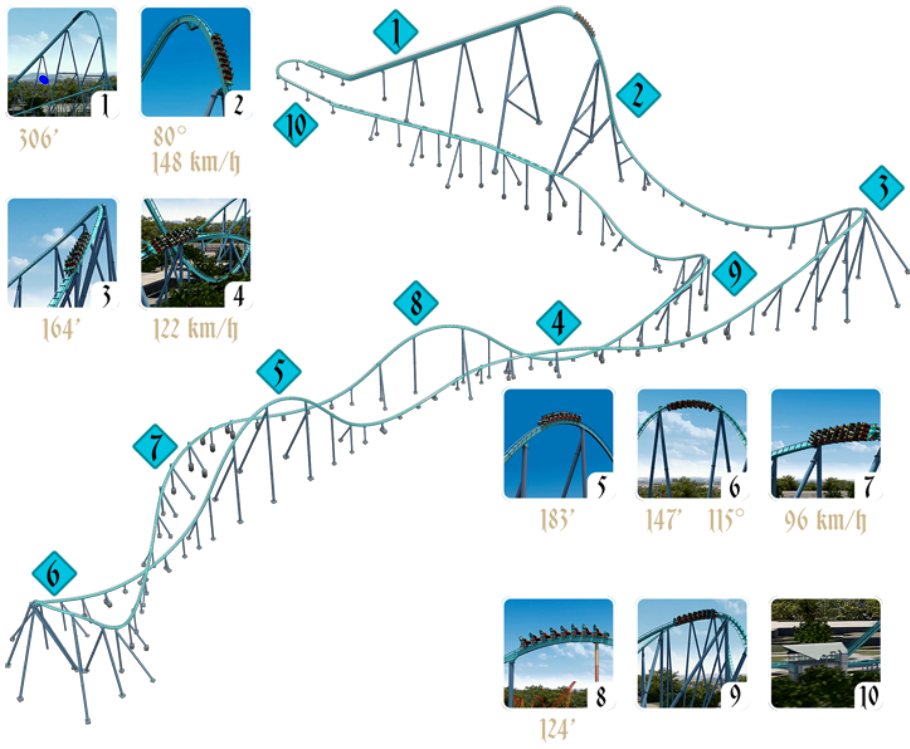
Energy can neither be created nor destroyed.

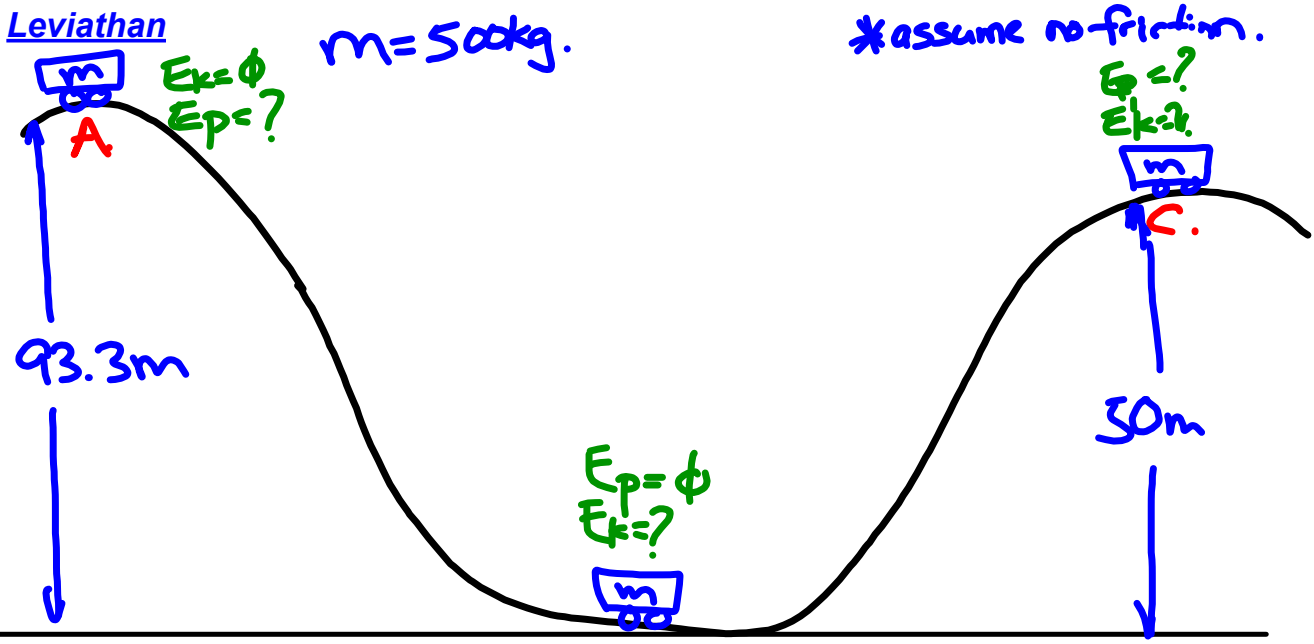
It can only change forms or be transferred from one object to another.

$$E_{Ti} = E_{Tf}$$

total initial energy = total final energy

Leviathan





A.

$$E_p = 457000$$

$$E_k = \phi$$

$$E_T = 457000$$

$$V = \phi \text{ m/s.}$$

B.

$$E_p = 0$$

$$E_k = 457000$$

$$E_T = 457000$$

$$V = \sqrt{\frac{2E_k}{m}}$$

$$= \sqrt{\frac{2 \times 457000}{500}}$$

$$= 42.8 \text{ m/s.}$$

$$\sim 154 \text{ km/hr.}$$

C

$500 \times 9.8 \times 50$

$$E_p = 245000$$

$$E_k = 212000$$

$$E_T = 457000$$

$$V = \sqrt{\frac{2E_k}{m}}$$

$$= \sqrt{\frac{2 \times 212000}{500}}$$

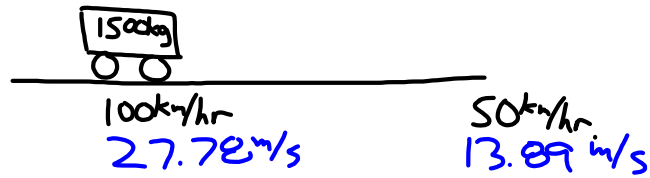
$$= 29.1 \text{ m/s.}$$

Warm Up / Practice Problems

1. Find the work required to slow down a 1,500kg automobile from 100 km/hr to 50 km/hr in a distance of 100m.

2. Find the work done by gravity when a 75 kg parachutist falls 750 metres.

1.

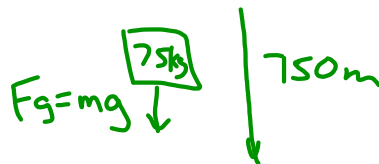


$$W = \Delta E = F \times d$$

$$\begin{aligned} W &= \Delta E \\ &= E_{k2} - E_{k1} \\ &= \frac{1}{2}(1500)(13.89)^2 \\ &\quad - \frac{1}{2}(1500)(27.78)^2 \\ &= -434000 \text{ J} \end{aligned}$$

$$\begin{aligned} F &= ma \\ \Delta d &= 100, v_1 = 27.78 \text{ m/s} \\ &\quad v_2 = 13.89 \text{ m/s} \\ a &= \frac{v_2^2 - v_1^2}{2\Delta d} \\ &= -2.89 \text{ m/s}^2 \\ F &= m a \\ &= 1500 \text{ kg} \times (-2.89 \text{ m/s}^2) \\ &= -4340 \text{ N} \\ W &= F \times d \\ &= -4340 \text{ N} \cdot 100 \text{ m} \\ &= -434000 \text{ J} \end{aligned}$$

#2



$$\begin{aligned} W &= F \times d \\ &= m \cdot g \cdot \Delta d \\ &= 75 \text{ kg} \times 9.8 \text{ m/s}^2 \times 750 \text{ m} \\ &= 551250 \text{ J} \\ &= 5.5 \times 10^5 \text{ J} \\ &\quad (550000 \text{ J}) \end{aligned}$$

What if all of the potential is not converted into kinetic energy?

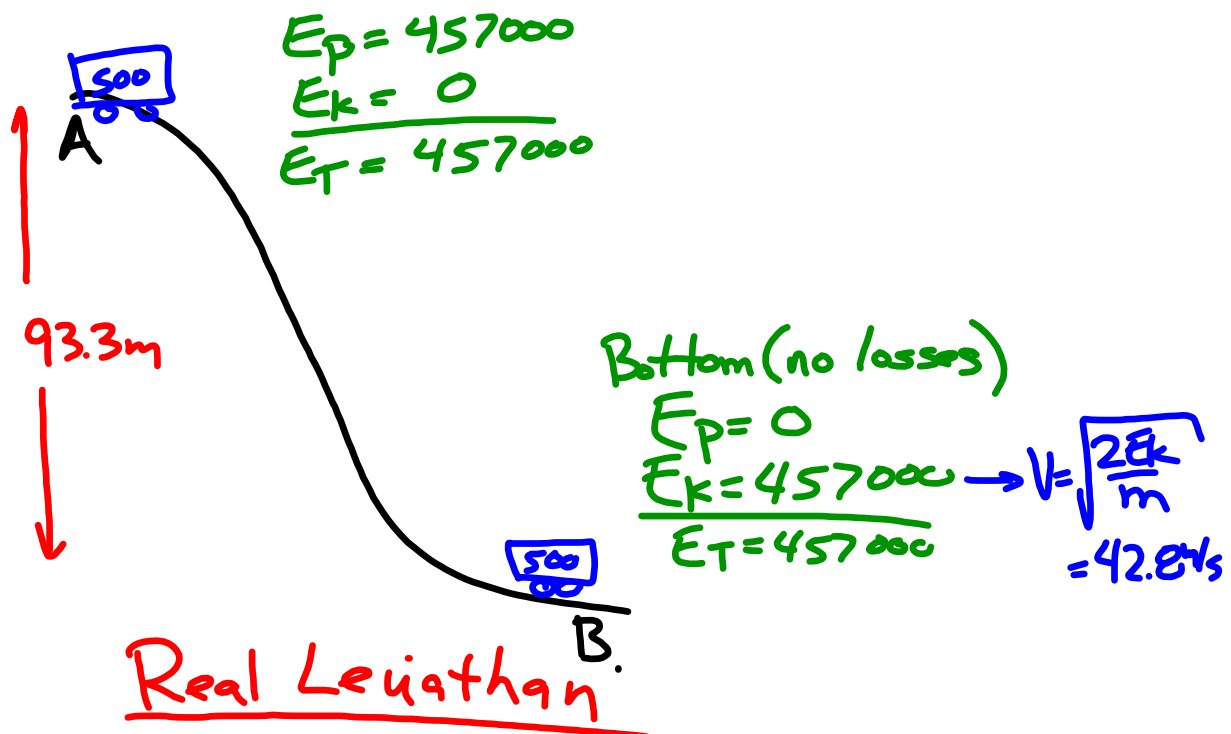
Where did it go?

Typically energy is "lost" to heat (friction), sound, light (radiant energy)

$$\text{efficiency} = \frac{\text{Useful energy out}}{\text{total input energy}} \times 100\%$$

$$\text{efficiency} = \frac{E_{\text{out}}}{E_{\text{in}}} \times 100\%$$

Example #1 : Calculate the efficiency of the Leviathan (a gravity machine that convert potential energy into kinetic energy)



Velocity = 148 km/hr
 = 41.1 m/s

$\Rightarrow E_k = \frac{1}{2} m v^2$
 $= 423000 \text{ J}$

energy

$E_p = 0$
 $E_k = 423000$

$E_{lost} = 34000 *$

$E_T = 457000$

efficiency = $\frac{E_{out}}{E_{in}} \times 100\%$
 $= \frac{423000}{457000} \times 100\%$
 $= 92.6\%$

Example #2 - Efficiency of a light bulb

An incandescent lamp requires $1.0 \times 10^4 \text{ J}$ of electrical energy to create 400 J of useful (i.e. visible) light energy. What is the efficiency of the incandescent bulb?

Where did the lost energy go



conversion from
electrical energy to light energy.

$$\text{efficiency} = \frac{E_{\text{out}}}{E_{\text{in}}} \times 100\% = \frac{400 \text{ J}}{1.0 \times 10^4 \text{ J}} \times 100\% = 4\%$$

lost to heat,
lost to other forms of radiation.

Power

Power is the rate of energy transformation or doing work.

$$\text{Power} = \frac{\text{change of energy}}{\text{time}}$$

$$= \frac{\text{work}}{\text{time}}$$

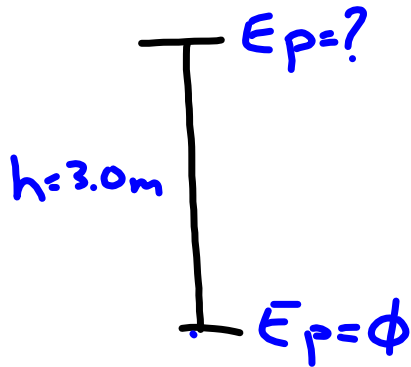
$$P = \frac{\Delta E}{t}$$

power can be energy consumed or generated.

$$\begin{aligned} \text{units} &= \frac{\text{Joules}}{\text{sec}} \\ &= \text{Watt} \end{aligned}$$

Examples:

1. A 75 kg man climbs up a 3.0 m ladder in 4.7s. What power must he generate to make this climb?



$$P = \frac{\Delta E}{t}$$

$\Delta E = \text{gain in potential}$

$$P = \frac{\text{gain in potential}}{t}$$

$$= \frac{mgh}{t}$$

$$= \frac{(75\text{ kg})(9.8\text{ m/s}^2)(3.0\text{ m})}{4.7\text{ s}}$$

$$= 470 \frac{\text{kg m}^2/\text{s}^2}{\text{s}}$$

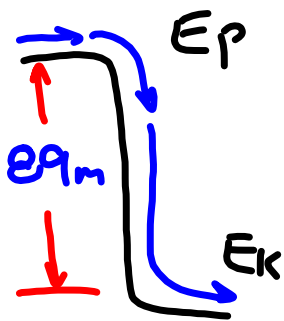
$$= 470 \text{ kg m}^2/\text{s}^3$$

$$P = 470 \text{ Watts}$$

2a. The Sir Adam Beck II generating station in Niagara Falls has a capacity of 1500MW.

The water driving the turbines drops 89 metres.

How many kg of water are needed per second, to generate 1500MW of power?



k → 1,000
M → 1,000,000

$$\begin{aligned} 1500\text{MW} \\ &= 1500\,000\,000\text{W} \\ &= 1.5 \times 10^9\text{W} \end{aligned}$$

generates 1.5×10^9 Joules of energy every second.

→ converting E_p of water to E_k

∴ need $1.5 \times 10^9\text{J}$ of E_p in the water.

$$E_p = mgh$$

$$m = \frac{E_p}{gh}$$

$$= \frac{1.5 \times 10^9\text{J}}{(9.8\text{m/s}^2)(89\text{m})}$$

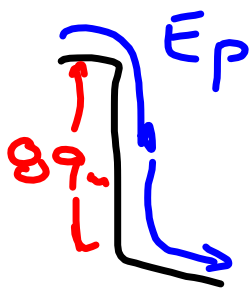
$$= 1.7 \times 10^6\text{kg}$$

$$= 1.7 \times 10^6\text{ litres}$$

2b. The actual volume of water available to the plant is 2.2×10^6 kg per second. What is the efficiency of the power plant in converting the potential energy of the water into electrical energy

$$\text{Power output} = 1.5 \times 10^9 \text{ W}$$

Available Power = ?



$$\begin{aligned} E_p &= mgh \\ &= (2.2 \times 10^6 \text{ kg})(9.8)(89 \text{ m}) \\ &= 1.92 \times 10^9 \text{ J} \\ &\text{energy second} \\ \text{or} &= 1.92 \times 10^9 \text{ W} \end{aligned}$$

$$\begin{aligned} E_{\text{eff}} &= \frac{P_{\text{out}}}{P_{\text{in}}} \times 100 \quad \left(\frac{E_{\text{out}}}{E_{\text{in}}} \times 100 \right) \\ &= \frac{1.5 \times 10^9 \text{ W}}{1.92 \times 10^9 \text{ W}} \times 100 \\ &= 78\% \end{aligned}$$

Home use of Power

Energy consumption in the house is measured in kWh's.

how many Joules in a kWh?

$$\begin{aligned}
 \underline{1 \text{ kWh}} &= 1000 \text{ W} \cdot \text{h} \\
 &= 1000 \text{ J/s} \cdot \text{h} \\
 &= 1000 \text{ J/s} \cdot (3600 \text{ s}) \\
 &= 3,600,000 \text{ J}
 \end{aligned}$$

Example

Calculate the cost to operate a 60 Watt light bulb for eight hours, if the cost per kWh is 7.2 cents.

1. calculate total energy used in Joules.
2. multiply by rate to get cost.

$$P = \frac{E}{t} \quad 1. \quad E = P \times t = 60 \text{ J/s} \times (8 \times 60 \times 60) = 1,728,000 \text{ J}$$

$$\begin{aligned}
 2. \quad \text{Cost} &= \text{energy used} \times \text{rate} \\
 &= (1,728,000 \text{ J}) \left(\frac{7.2 \text{¢}}{3,600,000 \text{ J}} \right) \\
 &= 3.5 \text{¢}
 \end{aligned}$$

Text Book Practice

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plus handouts - see website