Unit 3 : Energy Work & Power Review

Kou Tonics
Work
Mechanical Energy
Power
Nuclear Energy
Efficiency
Word Problems
Work and Mechanical Energy
Example : Calculate the energy required to lift a 505 kg roller coaster to the top of a 89m hill (assuming no
friction). What would the kinetic energy be at the bottom of the first hill if the drop was 75m (again assuming
no friction)?
Power
Example : If it took 1.5 minutes to get to the top of the hill, what power was used to move the coaster to the
top?
<u>Efficiency</u>
Example : If the actual energy used to get to the top of the hill was 5.0x10 ⁵ J (due to losses due to friction),
calculate the efficiency of the coaster lift mechanism.
<u>Thermal Energy</u>
Example : If 3.0 kg of methyl alcohol at 50°C is added to 2.0 kg of water at 20°C, what is the equilibrium
temperature (methyl alcohol c=2,460 J/kg°C, water c=4,180 J/kg°C)? Which substance gained or lost the most
energy?
Other stuff to know @
Convert from kWh to J
Convert from g to kg
Freezing and Boiling Points of Water, density of water
Units (MKS)
Review Mechanical Energy Quiz

Practice Problems

See back of this page and the following text book problems : page 186 #4, 8, 9, 11, 12, 18, 25, 27, 28, 29

Unit #3 Energy Work and Power Review Questions

#1. A student uses a force of 10.0 N to push her physics textbook a distance of 0.50 m across a table. Calculate the amount of work she does on the book.

#2. A wrestler lifts his opponent of mass 110 kg to a height of 2.8 m above the floor of the wrestling ring. How much work did the wrestler do on his opponent?

#3. A child pulls his wagon a distance of 15 m across the garden while applying a force of 160 N on the wagon's handle. How much work did the child do on the wagon?

#4. What is the kinetic energy of a 1000 kg car travelling at 25 m/s (assume 2 sig digits)?

#5. A man does 50 J of work to push a 3.0 kg shopping cart out of his way. If the cart started from rest, what was its speed immediately after this push? (assume there is no friction and that there are 2 sig digits)

#6. A 65 kg diver is standing on a diving platform 10.0 m above the water below.a. What is the gravitational potential energy of the diver with respect to the water's surface?b. What is her speed just as she strikes the water?

#7. A student pours 1.0 kg of water at 7° C into a kettle. Calculate the amount of heat energy needed to bring this water to a boil. The specific heat capacity for water is 4180 J/kg°C.

#8. A block made of an unknown substance has a mass of 1.5 kg and a temperature of 15° C. When 2.6 x 10^{4} J of heat is added to the block, its temperature rises to 90° C. Calculate the specific heat capacity of the unknown substance.

#9. Calculate the amount of heat energy required to melt a 10.0 kg block of ice at 0° C into liquid water at 0° C. The latent heat of fusion for water is 3.34×10^5 J/kg.

#10. What is the energy required to raise the temperature of 275 g of ice at -6.0°C to water at 78°C. The specific heat capacity of ice is 2100 J/kg °C, specific heat capacity of water is 4180 J/kg °C, latent heat of fusion of water is $3.34 \times 10^5 \text{ J/kg}$, latent heat of vaporization of water is $2.26 \times 10^6 \text{ J/kg}$.

#11. An elevator motor lifts the elevator full of passengers a height of 12 m in 30 s. If the mass of the elevator and passengers is 1000 kg, what is the power developed by the motor (assume 2 sig digs)?

#12 What is the equilibrium temperature when 125 grams of silver at 97°C are added to 2.5 litres of water at 21 °C? The specific heat capacity of water is 4180 J/kg °C and the specific heat capacity of silver is 240 J/kg °C.

#13. The efficiency of an incandescent light bulb is only 4.0%. Calculate the amount of electric energy that must be input to the bulb in order to get 100 J of light energy from the bulb.

Answers:

1. W = 5.0 J, **2.** $W = 3.0 \times 10^3 \text{ J}$, **3.** $W = 2.4 \times 10^3 \text{ J}$, **4.** $Ek = 3.1 \times 10^5 \text{ J}$, **5.** $v_2 = 5.8 \text{ m/s}$ **6a.** $Ep = 6.4 \times 10^3 \text{ J}$, **6a.** v = 14 m/s, **7.** $Q = 3.9 \times 10^5 \text{ J}$, **8.** $c = 231 \text{ J/kg} \cdot {}^\circ\text{C}$, **9.** $Q = 3.34 \times 10^6 \text{ J}$, **10.** $Q = 1.9 \times 10^5 \text{ J}$, **11.** P = 3900 W, **12.** $T = 21.2 \, {}^\circ\text{C}$, **13.** Ein = 2500 J