
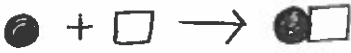
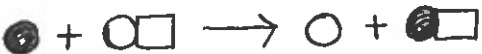



hydrogen + oxygen → water	1 reactant → 2 products
$AB \rightarrow A + B$	$AlCl_3 \rightarrow Al + 3Cl$
$2Na + Cl_2 \rightarrow 2NaCl$	
magnesium oxide → magnesium + oxygen	2 reactants → 1 product
water → hydrogen + oxygen	$A + B \rightarrow AB$
$3Cl_2 + 2Al \rightarrow 2AlCl_3$	magnesium + oxygen → magnesium oxide
	$2NaCl \rightarrow 2Na + Cl_2$
$3AgNO_3 + H_3PO_4 \rightarrow 3HNO_3 + Ag_3PO_4$	$2Al + 3ZnCl_2 \rightarrow 3Zn + 2AlCl_3$
element + compound → different element + compound	
$A + BC \rightarrow B + AC$	barium sulfide + potassium iodide → potassium sulfide + barium iodide
silver nitrate + sodium chloride → sodium nitrate + silver chloride	$ZnO + Ca \rightarrow CaO + Zn$
	$AB + CD \rightarrow AD + CB$
$ZnCl_2 + 2NH_4OH \rightarrow 2NH_4Cl + Zn(OH)_2$	different compound + compound → different compound + different compound
calcium fluoride + bromine → bromine + calcium fluoride	sodium chloride + sodium iodide → sodium chloride + sodium iodide