SNC 1DI

## **Density Problems**

Solve the following problems using the GRASSS method.

- 1. Calculate the mass of a liquid with a density of 3.2 g/mL and a volume of 25.0 mL.
- $D = \frac{1}{25}$ (25) 3.2 =  $\frac{1}{25}$  (25) D = 3.2g/mL15 = 250 mL M= 7 2. Calculate the density of a 500.0 g rectangular block with the following dimensions: length=8.0 cm, width=6.0 cm, height=5.0 cm.  $0 = \frac{1}{2}$  $= \frac{1}{2}$ M= 500 0 8 V=1×W+L = 8×6×5 = 743 cm<sup>7</sup> . the density of the block is Zglad D=7 3. Calculate the mass of a solid metal cylinder with a density of 2.6 g/cm<sup>3</sup>, a diameter D=2.6g/m of 1.8 cm, and a length of 4.0 cm. 0=5 VETTIL (11.35) 2.6 - M (11.707) = (3.14)( 9)2(4 m= (10,1736) (26) = 20.45 = 10.1736 cm3 M-D : the mail is 26 g 4. An irregular object with a mass of 18.0 kg displaces 2.5 L of water when placed in a
  - large overflow container. Calculate the density of the object.

- 18 = 72

the density is 72 Kg/L

 $D = \frac{m}{2}$ 

- M = 18 lcg V = 2.5 LD = ?
- 5. A graduated cylinder has a mass of 80.0 g when empty. When 20.0 mL of water is added, the graduated cylinder has a mass of 100.0 g. If a stone is added to the graduated cylinder, the water level rises to 45.0 mL and the total mass is now 156.0 g. What is the density of the stone?

 $D = \frac{M}{V} = \frac{45-20}{25 mL}$ =  $\frac{50}{25}$ = 2.24M = 150 - 80 - 20= 50 gThe density is 2.24 g(mL D =?